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AP ® Biology 2010 Scoring

Guidelines ... An experiment was conducted to measure the reaction rate of the human salivary enzyme α -amylase. Ten mL of a concentrated starch

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solution and 1.0 mL of α -amylase solution were placed in a test tube. The test tube was inverted several times to mix the solution and then incubated at

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College Board Chromatography Lab

see "The Free-Response Sections
□ Student Presentation" in the AP
Physics Course Description. 4.
The scoring guidelines typically
show numerical results using the
value $g = 9.8 \text{ m s}^2$, but use of 10
 m s^2 is of course also acceptable.

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Solutions usually show numerical answers using both values when they are significantly different. 5.

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Lab 10: Determining Mole Ratios.
Experiment 12: Some Descriptive
Chemistry. Experiment 13:
Classification of Chemical
Substances. AP Chem Lab Book
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25 10. Mgd M M Mqu $u = + = u. 2.$
 $uq == (10 \ 7 \ \sin \ 10 \ 7 \ 9.8 \ \text{m s} \ 4.0$

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$m \sin 30^\circ) g d () (2) () (\square)$ For a correct numerical answer . 1 point
. $u = 5.3 \text{ m s}$. Alternate solution (kinematics method) Alternate points . For determination of the linear acceleration in terms of mass and frictional force : 1 point
. $a_{FM} = 52 f$ from work shown in

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part (b) Chromatography Lab

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the linear acceleration in terms of
mass and frictional force : 1 point
. $a_{FM} = 52 f$ from work shown in
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BIO LAB 10 Circulation, but could
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questions. I just really want to

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make sure that my answers are correct. Here are the questions!
Questions Part A and B 1. Q.
Explain why blood pressure and heart rate differ when measured in a reclining position and in a standing position. A 2.

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AP® CHEMISTRY 2010 SCORING
GUIDELINES Question 3 (9 points)
 $8 \text{H}^+(\text{aq}) + 4 \text{Cl}^-(\text{aq}) + \text{MnO}_4^- (\text{aq}) \rightarrow 2 \text{Cl}_2 (\text{g}) + \text{Mn}^{3+} (\text{aq}) + 4 \text{H}_2\text{O} (\text{l})$
 $\text{Cl}_2 (\text{g})$ can be

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generated in the laboratory by reacting potassium permanganate with an acidified solution of sodium chloride. The net-ionic equation for the reaction is given above. (a) A 25.00 mL sample of 0.250 M NaCl reacts completely with excess KMnO_4

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AP CHEMISTRY 2010 SCORING GUIDELINES

Score: 10. The response earned 1 point in part (a) for correctly plotting the means on a bar graph. The response earned 1

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point in part (a) for appropriate labels, units, and scaling. The response earned 1 point in part (a) for correctly plotting the error bars.

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Samples Chromatography Lab

Briefly explain your answer. | An unknown solid has a melting point of 112-114°C. Based on data from Table 1 above and the mixed melting point technique of this experiment, you can narrow down the possibilities to three

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Compounds. List the compounds
b) A mixed melting point of the
unknown with acetanilide gives a
melting point of 113-114°C.

Melting Point Experiment
Experiment #10 - Melting ...

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1. To observe the reaction to be studied, transfer 10-mL of 1.5% (0.44 M) H₂O₂ into a 50-mL glass beaker and add 1 mL of the freshly made catalase solution. The bubbles coming from the

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reaction mixture are O_2 , which results from the breakdown of H_2O_2 by catalase. Be sure to keep the freshly made catalase solution on ice at all times.

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AP COVID-19 UPDATES. ... ask you to describe the results you would expect from an experiment. A good answer here will provide a rich, detailed account of the results you anticipate. A question that asks you to discuss a topic is

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Asking you for something broader than a mere description. A discussion is more like a conversation about ideas, and ...

AP Biology: How to Approach Free-Response Questions ...

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AP CHEMISTRY BY SATELLITE 9 - 7

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INSTRUCTOR LAB MANUAL

EXPERIMENT 9: SOLUBILITIES Pre-lab Questions: Answer these questions before coming to class. They introduce you to several important ideas that you will use in this experiment. You must turn-in this exercise before you will be

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allowed to begin the experiment.

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